TABLE VII

CONSTANTS<sup>a</sup> IN Eq. (4) FOR MOLAR VOLUMES OF FLUID ALONG THE MELTING CURVE

Fluid	a'	b'	e'	d'	$P_m$ range, $kg/cm^2$	rms dev., em³/mol
He <sup>4</sup> II He <sup>4</sup> I He <sup>3</sup>	0 14.854 1.075	-0.17145 $48.5273$ $51.1102$	$1 \\ -0.107253 \\ -0.161532$	27.570 $-10.0712$ $-3.2482$	26-30 35-3555 50-3555	0.0006 0.0097 0.0137

<sup>&</sup>lt;sup>a</sup> Pressure units in kg/cm<sup>2</sup> and volume units in cm<sup>3</sup>/mol.

## 2. Thermal expansion and compressibility of the fluid

The thermal expansion coefficient of fluid He³ along the melting curve exhibits a maximum in the vicinity of the triple point, as shown in Figs. 3 and 4. The maximum is broad compared to that for He⁴ and is less than one-half as large. In general, one expects  $\alpha$  to increase with T and decrease with P; however, along the melting curve the "normal" behavior of  $\alpha_f$  increasing with decreasing  $P_m$  and  $T_m$  indicates that  $P_m$  changes overcome  $T_m$  changes. For He⁴ the maximum in  $\alpha_f$  appears to be a direct consequence of the  $\lambda$ -transition. In He³ the nuclear spin part of  $\alpha_f$  becomes more negative at lower T, according to Goldstein (25), and it apparently overcomes the "normal" behavior of the nonspin part of  $\alpha_f$ .

From values of  $\alpha_f$  and  $\beta_f$  in Fig. 4, it is possible to compute the following thermodynamic quantities for fluid He<sup>3</sup> along the melting curve:

$$(\partial P/\partial T)_{V} = \alpha_{f}/\beta_{f}; \qquad (5)$$

and

$$(C_P - C_V) = TV_f \alpha_f^2 \beta_f. \tag{6}$$

These quantities are shown as the curves in Fig. 10. Neither curve exhibits a maximum over the range studied. The plot of  $(C_F - C_V)$  versus  $P_m$  is linear below 180 kg/cm<sup>2</sup> and extrapolates to zero at  $P_m = 47$  kg/cm<sup>2</sup>. This extrapolation gives a good determination of the point where  $\alpha_f$  goes through zero on the melting curve.

The pressure-temperature locus of  $\alpha_f = 0$  in the fluid domain is shown in Figs. 5 and 9. For completeness, the point of Taylor and Kerr (26) on the vaporization curve has been included. The points represented by open circles were obtained by extrapolation to zero of a series of  $\alpha_f$  values measured at constant pressure and various temperatures. This could be done reliably because the slopes were quite constant. Extrapolations were made below about 1.4°K, the

<sup>&</sup>lt;sup>3</sup> Lee et al., (27) also reported a density maximum at approximately 0.5°K, presumably at saturation.